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Article

Influence of Power Pulse Parameters on the Microstructure and Properties of the AlCrN Coatings by a Modulated Pulsed Power Magnetron Sputtering

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Abstract: In this study, AlCrN coatings were deposited using modulated pulsed power magnetron sputtering (MPPMS) with different power pulse parameters by varying modulated pulsed power (MPP) charge voltages (350 to 550 V). The influence of power pulse parameters on the microstructure, mechanical properties and thermal stability of the coatings was investigated. The results indicated that all the AlCrN coatings exhibited a dense columnar microstructure. Higher charge voltage could facilitate a denser coating microstructure. As the charge voltage increased up to 450 V or higher, the microvoids along the column boundaries disappeared and the coatings became fully dense. The main phase in the AlCrN coatings was the c-(Al, Cr)N solid solution phase with NaCl-type phase structure. A diffraction peak of the h-AlN phase was detected at a 2θ of around 33° , when the charge voltage was higher than 500 V. The hardness of the AlCrN coatings varied as a function of charge voltage. The maximum value of the hardness (30.8 GPa) was obtained at 450 V. All the coatings showed good thermal stability and maintained their structure and mechanical properties unchanged up to 800 °C during vacuum annealing. However, further increasing the annealing temperature to 1000 °C resulted in apparent change in the microstructure and decrease in the hardness. The charge voltages also showed a significant influence on the high-temperature tribological behavior of the coatings. The coating deposited at the charge voltage of 550 V exhibited excellent tribological properties with a low friction coefficient.

Keywords: modulated pulsed power magnetron sputtering; power pulse parameter; AlCrN coating; microstructure; tribological behavior

1. Introduction

In the last few decades, both high-power impulse magnetron sputtering (HIPIMS) and high-power pulsed magnetron sputtering (HPPMS) technologies have attracted significant attention from researchers and industry [1]. The HIPIMS technique applies short power pulses, with power densities of up to several kilowatts per square centimeter, on to the cathode target, resulting in high ionization of the deposition species. This technique improves the coating structure, improves certain mechanical properties such as hardness and adhesion strength between the coating and the substrate, and can be used to obtain a uniform deposition on workpieces with a complex shape [2–4]. However, some drawbacks such as a low deposition rate and charge instability due to peak voltages that are too high limit the application of the HIPIMS [5].

As an alternative to the HIPIMS technique, the modulated pulsed power magnetron sputtering (MPPMS) technique has relatively lower peak power compared to HIPIMS (several hundred W/cm² as compared to several kW/cm² in conventional HIPIMS) and longer pulse lengths (up to 3 ms as compared to 5–150 µs for conventional HIPIMS) [5]. In addition, MPPMS possesses two-stage pulses. In the first step, called the weak ionization stage, a low voltage is applied to ignite the plasma. Then the pulse is altered during the strongly ionized stage, to have high voltage, current and power. By using this kind of technique, the stability and control ability of high-power impulse magnetron sputtering are both further improved [5–14].

Using the MPPMS technique, Lin et al. [5–8], Li et al. [9], Papa et al. [10], Myers et al. [11–13], and Wang et al. [14] investigated the fabrication of metallic coatings (Cr, Ta), transition metal nitride coatings (CrN, Cr₂N, TiN), multilayer/super-lattice coatings (CrN/AlN), oxide coatings (TiO₂), and so forth. By varying the shape, width, and frequency of the power pulses, strong ionization and good coating properties can be obtained with the MPPMS technique. However, it has been noticed that most of the publications related to MPPMS are laboratory experiments, whereas the employed cathodes were mostly pure metals. In particular, more attentions need to be paid to the MPPMS deposition of multi-component coatings by alloy targets as most industrial hard coatings are multi-component coatings. In this article, we will report the fabrication of AlCrN coatings using an industrial MPPMS coater.

AlCrN and AlCrN-based coatings have been studied by researchers and industries due to their excellent tribological properties at elevated temperatures [15–17]. They have been extensively applied on to cutting tools, molds, and special components for antiwear applications [15,18]. Currently, most of the AlCrN coatings are deposited using arc ion plating or traditional DC/pulse magnetron sputtering (DCMS/PMS) techniques. However, the coatings that result from arc ion plating deposition present many macroparticles on the coating surface, resulting in high friction coefficient and wear rate [19,20]. The coatings deposited using magnetron sputtering usually exhibited a coarse columnar microstructure and low adhesion due to the low ionization rate of the magnetron sputtering [21]. Recently, the HIPIMS technique has been used to prepare AlCrN coatings. It was found that the AlCrN coating deposited by HIPIMS showed a very dense microstructure and a smooth surface, and thus had excellent mechanical properties [22,23]. Compared to HIPIMS, the MPPMS technique possesses the great advantages of better stability, control ability, and a higher deposition rate, all of which are beneficial for industrial production. However, AlCrN coatings synthesized using the MPPMS technique have rarely been reported. The influence of the pulse parameters of MPP, which are editable, on the microstructure and properties of AlCrN coatings is worth investigating.

Thus, in this paper, we fabricated the AlCrN coatings using a MPPMS technique in an industrial physical vapor deposition (PVD) coater. To investigate the influence of MPPMS plasma on the growth and properties of the coatings, we varied the charge voltage to obtain different pulses with various peak voltages and peak currents. The microstructure, mechanical properties, thermal stability, and tribological properties of the coatings were investigated.

2. Materials and Methods

An industrial scale coating unit, the Flexi coat 1000, with a chamber volume of 1 m³ manufactured by IHI Hauzer Techno Coating B.V. (Venlo, The Netherlands) was used for coating deposition. The coating unit was equipped with four sputtering cathodes, two of them with MPPMS power supply, and two with DC sputtering power. In this study, we used the two facing MPPMS sputtering cathodes with AlCr targets (99.99 at %, Al₆₀Cr₄₀, at %). The substrates in this study included polished WC-Co cemented carbide coupons (30 × 30 × 8 mm³, R_a = 34.6 nm) and N-type (100)-orientated Si wafers (R_a = 5 nm). The substrates were ultrasonically cleaned in acetone and alcohol for 15 min each, then dried and fixed in the coating chamber before coating deposition. After the coating chamber was evacuated to a vacuum of less than 5.0 × 10^{−3} Pa, Ar gas at a flow rate of 130 sccm was introduced into the chamber. Presputtering of the target and plasma cleaning of the substrates with a plasma

source were conducted; both procedures aimed to remove the impurities and absorb the gaseous species. Then coating deposition began with the two AlCr MPPMS targets sputtering in an Ar + N₂ gas environment. The Ar and N₂ gas flow rates were set at 200 and 300 sccm, respectively. The deposition temperature was kept at 450 °C. Figure 1 schematically shows the waveform and parameters of the MPP power pulses applied on to the sputtering targets. For the MPP power pulses, a repeated two-stage pulse (τ_{on} in Figure 1) was utilized. In the first stage, referred to as the weak ionization stage (τ_w), a low voltage is applied to ignite the plasma. Then the pulse proceeds to the strongly ionized stage which has a high voltage (τ_s). Table 1 lists the MPP power parameters used in this study and the resulting coating characteristics. During coating deposition, MPP power pulses with a fixed average power P_a , constant pulse on time τ_{on} , and constant τ_w/τ_s ratio were utilized. By increasing the charge voltage of the MPP power, the peak voltage, peak current, and peak power of the power pulses increased (Table 1). To keep fixed the average power, the frequency of the power pulses changed accordingly, as shown in Table 1.

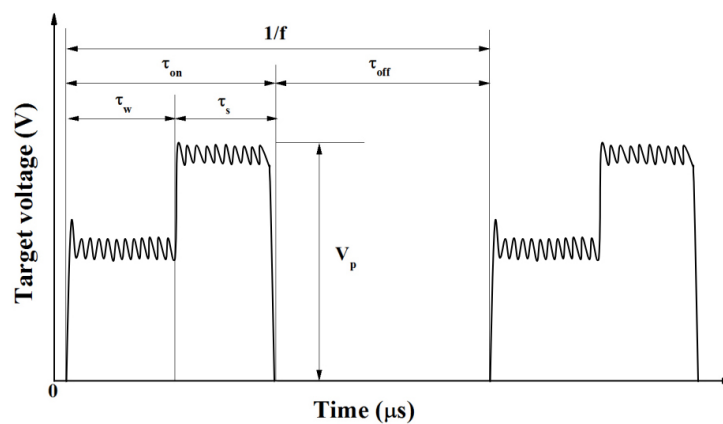


Figure 1. Typical waveform and parameters of the modulated pulsed power (MPP) power pulses. τ_{on} : the pulse-on time; τ_w : the weakly ionized pulse time; τ_s : the strongly ionized pulse time; f : the pulse frequency; P_a : the average target power; P_p : the peak power of the pulses; V_p : the peak voltage of the pulses; I_p : the peak current of the pulses.

Table 1. Deposition parameters using different modulated pulsed power magnetron sputtering (MPPMS) charge voltages in this study.

Sample	Charge Voltage V_a [V]	MPP Pulse Parameters							Coating Hardness [GPa]	Coating Thickness [μ m]
		τ_{on} [μ s]	τ_w/τ_s [μ s]	f [Hz]	P_a [Kw]	P_p [kW]	V_p [V]	I_p [A]		
C1	350	1500	800/700	203	10.0	56	433	130	24.9 \pm 7.3	3.3
C2	400	1500	800/700	120	10.0	95	492	192	26.6 \pm 5.9	2.9
C3	450	1500	800/700	87	10.0	134	552	243	30.8 \pm 5.5	3.0
C4	500	1500	800/700	68	10.0	158	622	253	26.5 \pm 5.4	3.5
C5	550	1500	800/700	45	10.0	197	691	287	20.6 \pm 6.6	3.9

The phase structure of the as-deposited coatings was characterized on an X-ray diffraction (XRD, Philips, Almelo, The Netherlands) machine in the $\theta/2\theta$ geometry using Cu K α radiation (40 kV and 40 mA). The morphologies and chemical compositions of the coatings were examined using a scanning electron microscope (SEM, JSM-6701F, JEOL, Tokyo, Japan) equipped with an energy-dispersive X-ray spectrometer. The EDX measurements were performed on a flat film surface (no tilt) with the following parameters: A 10 kV accelerating voltage and a fixed working distance of 4 mm. The probe current was automatically set and was recorded as around 0.16 nA. The hardness (H) and elastic modulus (E) were measured using a nanoindenter with a Berkovich diamond indenter (NHT², CSM, Peseux, Switzerland).

To investigate the thermal stability and the structural changes of the coatings at an elevated temperature, annealing experiments were conducted at 600 °C, 800 °C, and 1000 °C for 2 h in vacuum ($\leq 1 \times 10^{-2}$ Pa). The heating rate was kept at 5 °C/min. After thermal annealing, the vacuum chamber was air cooled to room temperature. The tribological properties were investigated using a high-temperature ball-on-disc tribometer (HT-1000, CSM, Peseux, Switzerland) at temperatures of 600 °C and 800 °C. Al₂O₃ balls with a 6 mm diameter were used as the counterparts. A normal load of 5 N was applied on the ball. The average sliding speed was 0.03 m/s for a fixed period of 50,000 rotating cycles. To measure the wear loss, the two-dimensional (2D) cross-sectional profile of the wear tracks were obtained using a Taylor Hobson CCI optical profiler. The wear rates ($W, \times 10^{-7} \text{ mm}^3/\text{N}\cdot\text{m}$) of the coatings were calculated with the following formula: $W = W_r/(L \times S_d) = S \times 2\pi R/(L \times 2\pi r \times n) = S/(L \times n)$, where W_r is the wear volumes of coating, L is the load, R is the radius of wear crack, n is the number of wear laps and S is the cross-sectional area of wear crack. The wear tracks were analyzed using the aforementioned SEM/EDX machine. X-ray photo electron spectroscopy (XPS, PHI-Quantum 2000, Eden Prairie, MN, USA) was performed to observe the bonding states of the tribo-oxides in the wear tracks. During XPS analysis, an Al K α X-ray source (13 kV and 15 mA) was used. The XPS spectra were obtained after removing the surface layer of the sample by in situ sputtering with Ar⁺ ions (3 keV) for 40 s. The scanning range for the full spectrum and individual spectrum was 100 and 30 eV, respectively. The scanning step for the full spectrum and individual spectrum was set at 1 and 0.1 eV, respectively. The individual spectra were calibrated using a carbon peak C 1s at 284.5 eV.

3. Results and Discussion

3.1. Chemical Composition, Morphology, and Phase Structure of the Coatings

Figure 2 shows the Cr, Al and N content (EDX results) in the AlCrN coatings as a function of the charge voltage of the AlCr targets. When the charge voltages were no higher than 450 V, the coatings were over-stoichiometric, with the N content being detected as almost constant: ~56 at %. The Al/Cr ratios were kept at around 1.3:1, which was lower than the ratio in the AlCr target (1.5:1). The decreased Al contents would be attributed to the lower sputtering rate of Al than Cr [24]. The N content decreased rapidly and stayed at ~52 at % when the charge voltage increased to 450 V or above. Accordingly, the Al content increased from ~25 at % to ~28 at % and the Cr content increased from ~19 at % to ~20 at %. As shown in Table 1, higher charge voltage of MPP power source corresponded to higher scales of peak voltage, peak current and increased peak power, which resulted in higher plasma density and stronger ionization of deposition species. The improved ionization would enhance the ion bombardment and resulted in more re-sputtering of nitrogen for its lower mass than the metal elements, which consequently led to lower N contents. Moreover, the formation of a non-conductive AlN film on the target surface has been reported to lead to target poisoning [25]. In the current work, the higher ionization of MPP deposition species at higher charge voltages is expected to reduce the target poisoning, resulting in higher sputtering rate of Al species from the target than that at lower charge voltages. Therefore, lower N content and higher Al content was observed when the charge voltage increased to higher than 450 V in Figure 2.

Figure 3 shows the XRD patterns of the AlCrN coatings deposited at various MPP charge voltages. When the charge voltage was lower than 500 V, the AlCrN coatings showed a typical B1-NaCl FCC structure with identifiable (111) and (200) diffraction peaks of the c-(Al, Cr) N phases. The (220) diffraction peak at 63–65° is clearly positioned due to its overlapping with a diffraction peak of the substrate. No diffraction peak corresponding to hexagonal h-AlN phase was observed. When the charge voltage reached 500 V, an h-AlN (100) diffraction peak at $2\theta \approx 33^\circ$ appeared. No diffraction peak corresponding to the hexagonal h-AlN phase was observed. When the charge voltage reached 500 V, an h-AlN (100) diffraction peak at $2\theta \approx 33^\circ$ appeared. When the charge voltage was further increased to 550 V, the h-AlN (100) diffraction peak becomes stronger and the h-AlN (002) diffraction peak also appears in the pattern. The results suggest that high voltage charge favors the formation

of h-AlN phase. As discussed above, an increase in the charge voltage corresponds to an increase in the peak power (Table 1) and a higher metal ionization ratio. With the application of bias voltage to the substrates, which was -100 V in this study, more metallic ions were attracted to the substrate with accelerated speeds. At higher energy, the atoms/ions prefer to occupy positions with the lowest free energy. Some of the Al-N bonds could agglomerate in the (Al, Cr)N solid solution and form the h-AlN phase. Therefore, the phase separation was more apparent in the coatings deposited at high charge voltages (500–550 V). Similar results were reported by Bagcivan et al. [26]. In their research, $(\text{Cr}_{1-x}\text{Al}_x)\text{N}$ coatings were deposited by DCMS, MF-MS (Middle Frequency Magnetron Sputtering), and HPPMS techniques, in which the Al/Cr ratios for initial h-AlN formation were compared. It was found that the formation of h-AlN phase in the HPPMS coatings was much easier than in other techniques. The Al/Cr ratio was high than 50:50 for HPPMS for the first appearance of h-AlN phase, while much higher Al/Cr ratio (77:23) was needed in the DCMS and MF-MS CrAlN coatings [26].

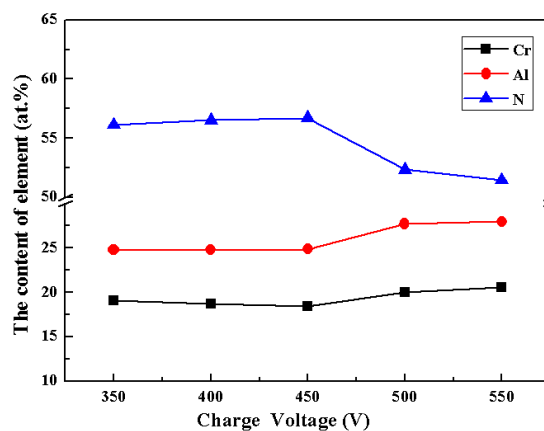


Figure 2. Chemical composition of AlCrN coatings as a function of charge voltage.

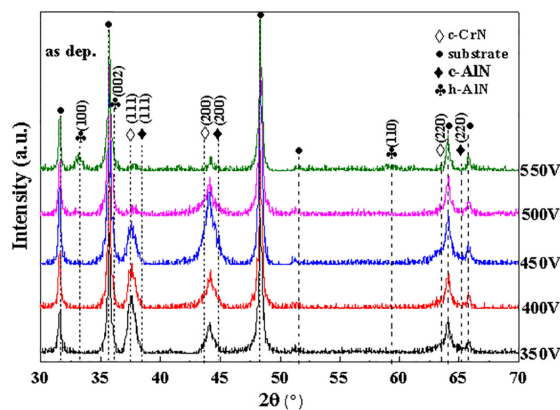


Figure 3. X-ray diffraction (XRD) patterns of AlCrN coatings deposited at various charge voltages.

As the charge voltage was increased from 350 to 450 V, it was found that the preferred orientation changed from (111) to (200). As reported by Paulitsch et al. [27] and Petrov et al. [28], the (200) orientation can be stabilized by a high ion to neutral ratio. A similar transition of the preferred growth orientation from (111) to (200) due to increased ion bombardment was also found in magnetron sputtered TiAlN/VN and CrN/NbN coatings [29,30]. The transitions occur through competitive grain growth. The increased ion bombardment favors renucleation and growth of (200) grains. In this study, a high charge voltage meant high ionization and a higher ion to atom ratio during the deposition process, which explains the preferred orientation evolution. It was also found that the diffraction peaks were broadened, which indicates decreased grain size. Further increases to the charge voltages induced

much lower diffraction peaks, indicating decreased crystallinity. To clarify this point, we estimated the average grain sizes of the c-(Al, Cr)N phases using a Williamson-Hall plot [31], as summarized in Figure 4. It can be seen that the grain size of the c-(Al, Cr)N phase initially decreased from ~22.8 nm at 350 V to ~8.0 nm at 450 V and then increased again to ~23.8 nm with a further increase in the charge voltage. As discussed before, the increase in the charge voltage corresponds to the increase in the peak power (Table 1) and higher metal ionization ratio. The increase in the impinging energy of the forming atoms/ions and enhanced ion bombardment favored grain refinement and a (200) preferred orientation at a charge voltage of 450 V. When the charge voltages was further increased, the thermal effect of the ion bombardment induced a slight grain growth, again (Figure 4).

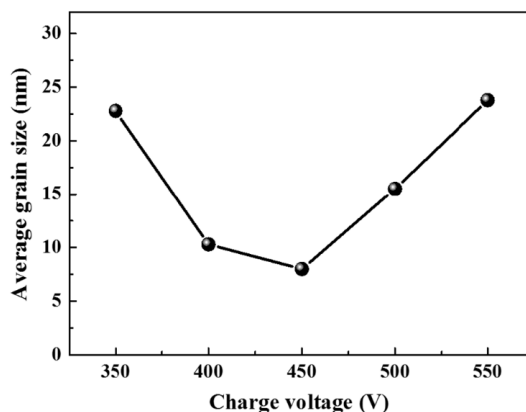


Figure 4. Average grain size of AlCrN coatings as a function of the charge voltage.

Figure 5 shows the surface and cross-sectional SEM images of the AlCrN coatings deposited at the charge voltages 350 V, 450 V, 500 V and 550 V. At 350 V, the coating exhibited an apparent columnar microstructure with columns that are ~0.2–0.3 μm in diameter. The columns are not closely packed together. Some penetrating voids can be detected between the columns (Figure 5a). When the charge voltage increased to 450 V, the voids along the column boundaries disappeared and the film became fully dense. When the charge voltage was further increased to 500 V and 550 V, coatings with featureless and glass-like microstructures were observed. As discussed above, an increased charge voltage corresponded to a higher ionization ratio, which would enhance ion bombardment and disrupt the local epitaxial growth of individual columns, resulting in renucleation of the grains and micro columns [32]. This can explain both the grain refinement and coating densification at higher charge voltages. The results are consistent with the observations of microstructure evolution reported in previous work [28,29].

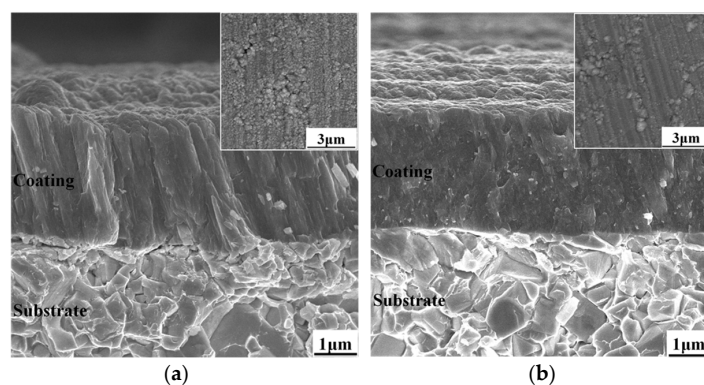


Figure 5. Cont.

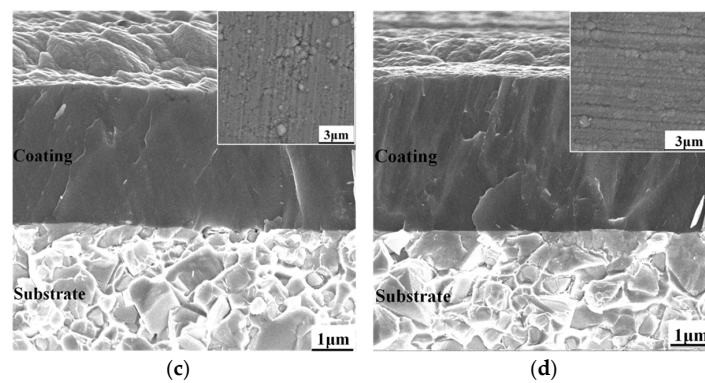


Figure 5. The surface (insets) and cross-sectional scanning electron microscopy (SEM) micrographs of AlCrN coatings deposited at different charge voltages: (a) 350 V; (b) 450 V; (c) 500 V; (d) 550 V.

3.2. Mechanical Properties of the Coatings

Figure 6 shows the hardness and elastic modulus values of the AlCrN coatings as a function of charge voltage. The values of coating hardness and elastic modulus in this study are comparable to the values found in other studies [22,26]. As the charge voltage increased, the hardness and elastic modulus of the AlCrN coatings initially increased from 24.4 GPa and 401.6 GPa at 350 V to maximum values of approximately 28.6 GPa and 439.5 GPa at a charge voltage of 450 V, respectively. As discussed above, the increase in the impinging energy of the forming atoms/ions and enhanced ion bombardment favored grain refinement and a (200) preferred orientation at a charge voltage of 450 V (Figures 3 and 4), as well as the microstructure densification (Figure 5). The enhanced hardness at a charge voltage of 450 V is believed to originate from the microstructure densification and grain refinement. A further increase in the charge voltage to 500 V and 550 V resulted in a decrease in the hardness and elastic modulus (Figure 6). The rebounding mechanical properties should be related to the grain growth and the formation of the softer h-AlN phase in the coatings (Figures 3 and 4). Similar tendencies were found in similar coatings accompanying the increases in ionization ratio and ion bombardment [22].

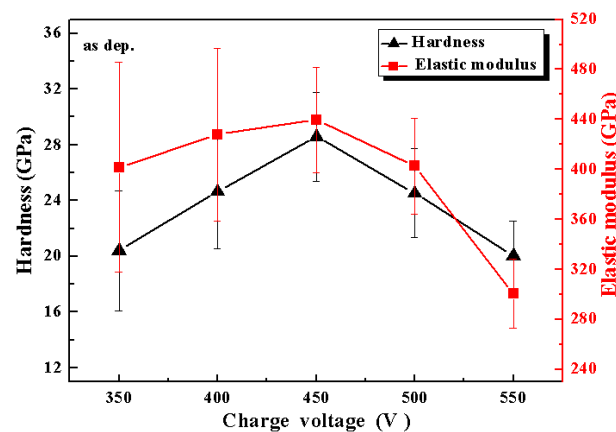


Figure 6. The hardness and elastic modulus of AlCrN coatings deposited at different charge voltages.

3.3. Thermal Stability of the Coatings

Figure 7a–c show the XRD patterns of the AlCrN coatings after vacuum annealing at 600 °C, 800 °C and 1000 °C, respectively. To clarify that the XRD peaks were changed by thermal annealing, we added the labeling of the unaltered XRD peaks as “+” in Figure 7. After being annealed at 600 °C (Figure 7a), all the AlCrN coatings exhibited similar diffraction patterns to those of the as-deposited coatings (Figure 3). In the XRD patterns from coatings annealed at 800 °C (Figure 7b), it is evident that

most of the diffraction curves are similar to those of the as-deposited and 600 °C annealed coatings, except that some weak peaks were identified as hexagonal h-AlN and h-Cr₂N phases within the coating deposited at a 550 V charge voltage. This suggests that most of the AlCrN coatings maintained thermal stability at temperatures below 800 °C. By further increasing the annealing temperature to 1000 °C (Figure 7c), diffraction peaks of the hexagonal h-AlN and h-Cr₂N phase were observed in most coatings. Because the coatings deposited by PVD process are thermodynamically meta-stable, the c-(Al, Cr)N solid solution phase containing high Al content is expected to undergo partial decomposition at high temperature to form thermodynamically stable phases of h-AlN and Cr₂N. It should be pointed out that even at 1000 °C, the fraction of decomposed c-(Al, Cr)N phase was very limited (Figure 7c). An advantage of the ternary Al-Cr-N system was reported to be the thermal stability of the cubic AlN phase even at very high Al-concentrations [2]. For example, Al concentrations of 60–75 at % have been reported in the stabilized cubic AlCrN phases [3,26].

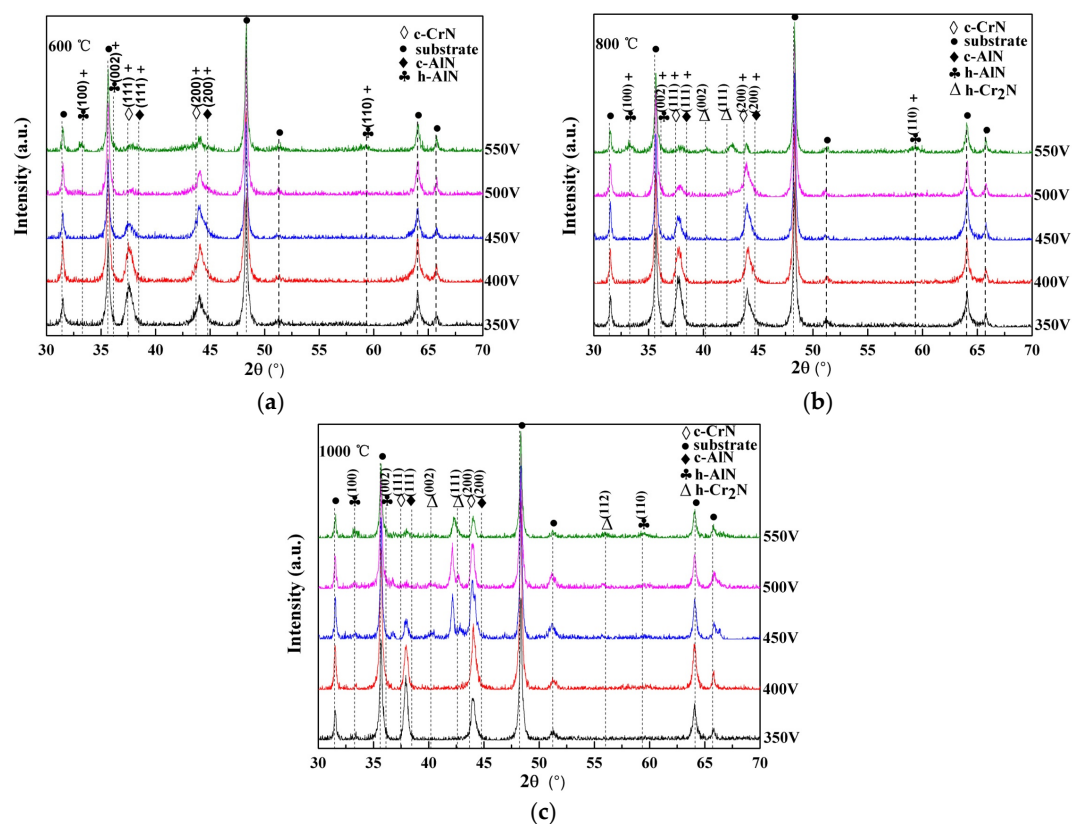


Figure 7. XRD patterns of AlCrN coatings deposited at different charge voltages after vacuum annealing for 2 h at (a) 600 °C, (b) 800 °C, and (c) 1000 °C. “+” denotes the XRD peaks unaltered by the thermal annealing.

Figure 8 shows the cross-sectional SEM micrographs of typical AlCrN coatings (deposited at 450 V and 500 V) after vacuum annealing. Corresponding to the XRD results in Figure 7a,b, no apparent microstructure change was observed in the coatings after vacuum annealing at 800 °C (comparing Figure 8a,c and Figure 5). However, after vacuum annealing at 1000 °C, granular grains were observed in the coatings deposited at both 450 V and at 500 V (Figure 8b,d), implying grain coarsening in the coatings. The coatings transformed from a very fine columnar microstructure to a granular crystal microstructure. These results are consistent with the XRD results in Figure 7c. Apparently narrower diffraction peaks of the c-(Al, Cr)N phase indicate improved crystallization. Both h-AlN and Cr₂N peaks were also observed. As discussed before, the coatings deposited using the PVD process are far from thermodynamic equilibrium. The coatings will try to approach thermodynamic equilibrium at a

high temperature. A decrease in defect density and grain growth during the thermal annealing process will be expected. Higher annealing temperatures favor restoration and grain coarsening process. Therefore, we observed different coating microstructures in Figure 8.

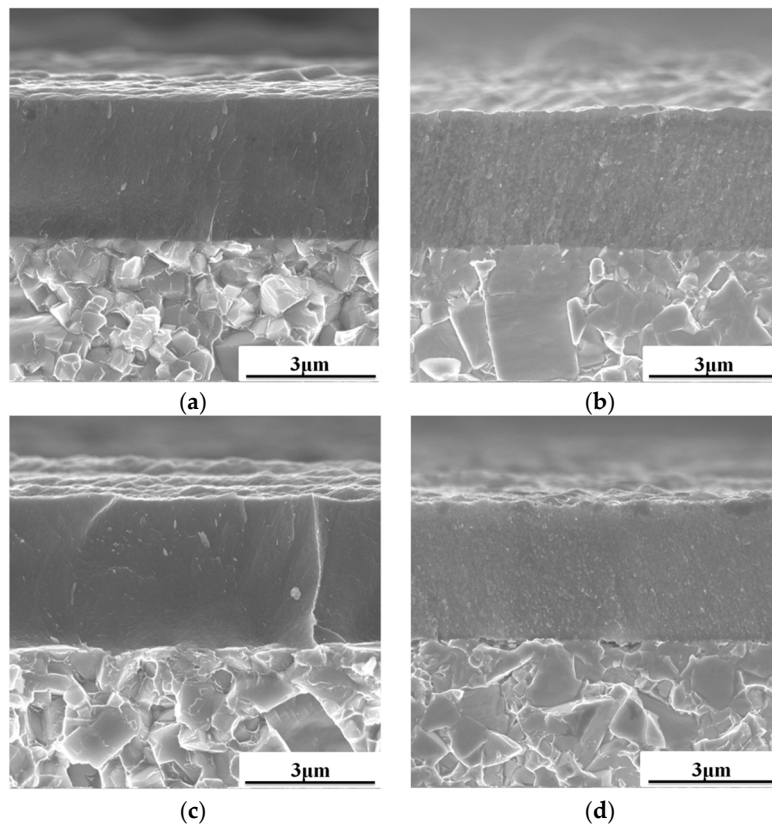


Figure 8. The cross-sectional SEM micrographs of the vacuum annealed AlCrN coatings deposited at the AlCr target charge voltage of (a,b) 450 V and (c,d) 500 V after vacuum annealing for 2 h at (a,c) 800 °C and (b,d) 1000 °C.

As shown in Figure 9, when the vacuum annealing temperature was below 800 °C, the hardness values were almost the same as those of the as-deposited coatings. This suggests that the AlCrN coatings maintained thermal stability at temperatures below 800 °C, which is consistent with both the XRD results (Figure 7) and the SEM results (Figure 8). Similar results were reported by Feng et al. [33] and Lin et al. [34]: Hardness of the AlCrN coatings drops substantially at temperatures higher than 700 °C. The formation of softer hcp-AlN and other new phases are the reason behind the sharp drop in hardness at high temperatures. After annealing at 1000 °C, significant drops in the hardness values were found for all the coatings. The grain growth and decomposition of the c-(Al, Cr)N solid solution into h-AlN and Cr₂N phases should explain the softening of the coatings at high temperatures. Therefore, the AlCrN coatings exhibited lower hardness after annealing at 1000 °C.

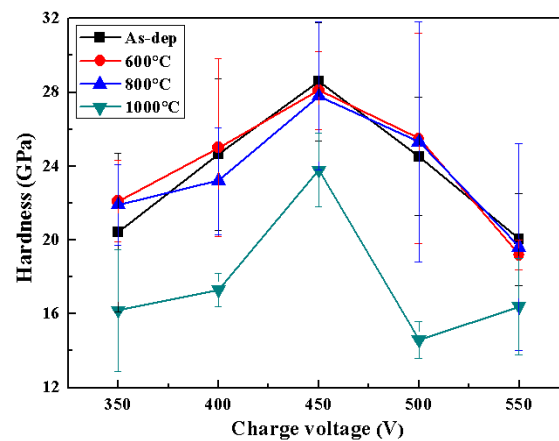


Figure 9. The evolution of the hardness values after vacuum annealing at 600, 800 and 1000 °C for 2 h for AlCrN coatings deposited at different charge voltages.

3.4. Tribological Properties of the Coatings

The friction coefficient curves of the AlCrN coating sliding against a Al_2O_3 ball at 600 °C and 800 °C are presented in Figure 10. It was found that the friction coefficient curves can be divided into two periods: the running-in period and the steady-state period. The friction coefficients of all the coatings experienced a rapid increase in the initial running-in period, which was considered a result of the transition in contact conditions from two-body abrasion to interfacial sliding. In a similar ball-on-disc dry sliding wear of a sputtered nitride coating, the running-in friction was found to be associated with a progressive transition of the sliding contact from two-body sliding to the formation of a tribofilm [35]. A relatively steady state was reached after some rotating laps. The friction coefficient curves for all the coatings were stable at 600 °C after around 500 laps except in the case of the coating deposited at a charge voltage of 550 V. At 800 °C, the friction coefficient curves of all the coatings decreased as the number of rotating laps increased, indicating a gradual formation of oxides, which were able to play the role of solid lubricants [36]. However, the curves for the AlCrN coating deposited at a charge voltage of 550 V exhibited a much lower friction coefficient than the other coatings at the test temperature of both 600 °C and 800 °C. This may be attributable to the lower coating hardness induced by the formation of the softer h-AlN phase in the coatings. There was also another possibility that a lubricated WO_3 tribo-film was formed by the tribo-oxidation of WC-Co substrates after the coatings wore out. This possibility will be discussed later.

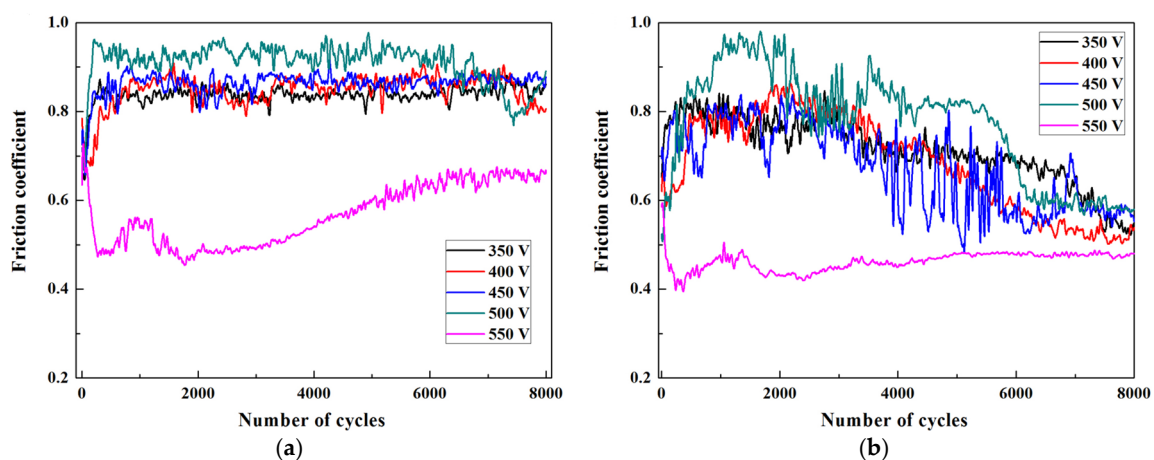


Figure 10. Friction coefficient curves of AlCrN coatings deposited at different charge voltages under (a) 600 °C and (b) 800 °C.

Figure 11 shows the 2D cross-sectional profiles of the wear tracks of the AlCrN coatings. It is observed that there were grooves with different depths on the wear tracks of the AlCrN coatings after wear tests. The depths of the wear tracks were different depending on the coating parameters and test temperatures. To clarify this point, the wear rates (W , $\times 10^{-7} \text{ mm}^3/\text{N}\cdot\text{m}$) of the coatings were calculated and are shown in Figure 12. It can be seen that the wear rates at 600 °C and 800 °C exhibited similar changes with the increase in charge voltage. The coatings experienced more severe wear at 800 °C than that at 600 °C. The wear rates are lower in the coatings that were deposited at 400 V and 450 V, compared to the other deposition parameters. The lowest wear rates of $7.7 \times 10^{-6} \text{ mm}^3/\text{N}\cdot\text{m}$ and $11.2 \times 10^{-6} \text{ mm}^3/\text{N}\cdot\text{m}$ were observed at 600 °C and 800 °C, respectively. The coatings deposited at charge voltages of 400 V and 450 V possessed higher hardness than the other coatings (Figure 6). This is likely to be the reason for their lower wear rates. At 800 °C, the oxidation of the coatings and the softening of the substrates should occur to a greater degree compared to the coatings tested at 600 °C. Therefore, these coatings experienced more severe wear compared to those tested at 600 °C.

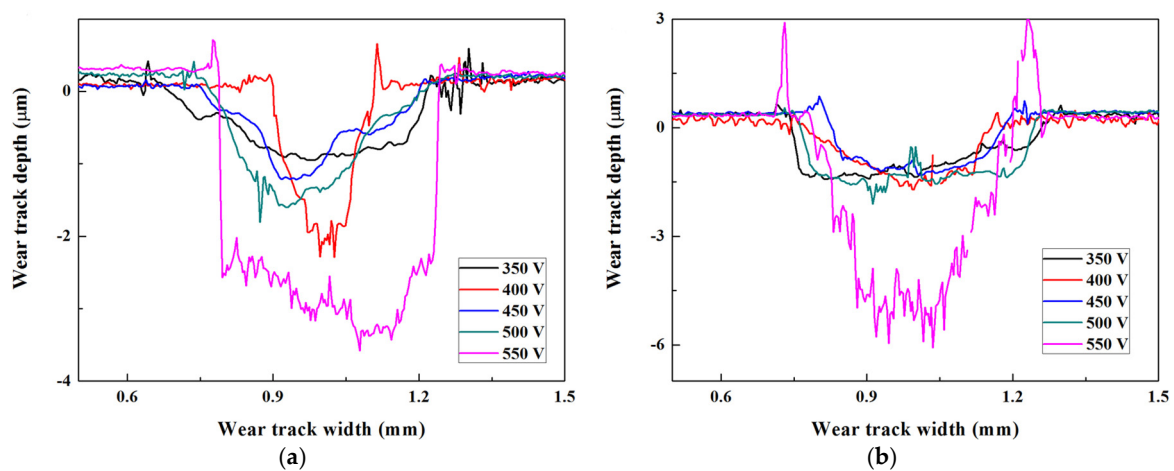


Figure 11. The 2D profile curves of the wear tracks for AlCrN coatings after wear tests under (a) 600 °C and (b) 800 °C.

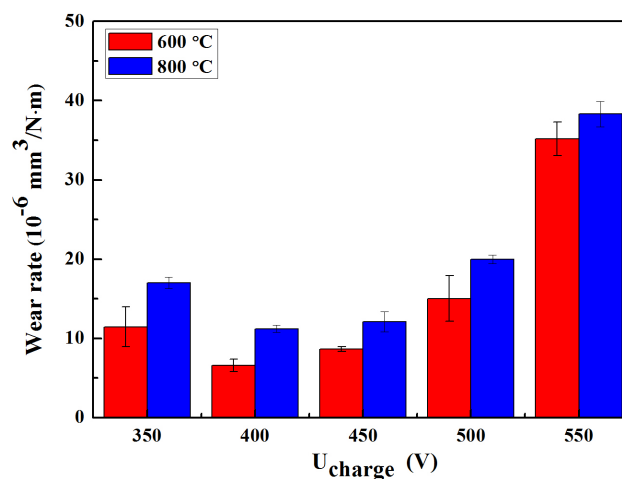


Figure 12. The wear rates of AlCrN coatings deposited at different charge voltages in the pin-on-disk wear tests at 600 °C and 800 °C.

To clarify the wear products on the AlCrN coatings, the wear tracks were analyzed using both SEM/EDX and XPS. Figure 13 shows the typical worn surface of the coating deposited at 450 V after wear tests at 800 °C. The wear tests of the other coatings deposited at 350–500 V are similar to this one. In Figure 13, it can be seen that the coating surface can be divided into an intact surface (A zone in Figure 13) and a worn surface (B zone in Figure 13). On both the intact coating surface and on the wear track, there exhibits an AlCrN based EDX spectrum, which suggests only mild wear by partial tribo-oxidation. To identify the oxide species on both the wear tracks, the specimens were analyzed further using XPS.

The results of the detailed XPS analysis in the wear tracks are shown in Figure 14. The Cr 2p peaks acquired in the 600 °C wear track fit well with chromium nitrides (Cr_2N 574.2 eV and CrN 575.3 eV), whereas the peaks shift to chromium oxide (Cr_2O_3 577.4 eV) in the 800 °C specimen. A similar shift from AlN (73.5 eV) to Al_2O_3 (75.2 eV) appeared in the Al 2p peak, as shown in Figure 14b. In Figure 14c,d, the O 1s peak is substantially higher and the N 1s peak has become marginal in the 800 °C wear track as compared to that in the 600 °C specimen. The XPS analyses confirmed that the coatings in the wear tracks after wear tests at 800 °C were more oxidized by tribo-oxidation than those after wear tests at 600 °C. More Al_2O_3 and Cr_2O_3 formed while less AlN and CrN species were left on the worn surface in 800 °C test. No other species were found. Therefore, the AlCrN coatings were subjected to oxidation into Al_2O_3 and Cr_2O_3 during high-temperature wear tests. Tribo-oxidation occurred similarly for the coatings deposited at various charge voltages and exhibited similar friction coefficients, except for the coatings deposited at 550 V.

The wear tracks on the AlCrN coatings deposited at 550 V are shown in Figure 15. Unlike the EDX analyses presented in Figure 13, these wear tracks show substantial amounts of W, Co and O, indicating the exposure of the WC based substrate and its tribo-oxidation. In addition, the depth of the wear tracks exceeds the thickness of the AlCrN coating (Figure 11), which indicates that the coating was totally worn out during the wear process. It has been reported that the tribo-oxidation of a WC substrate resulted in the formation of a lubricious Magnéli phase WO_3 [37], which resulted in the low friction coefficients seen in Figure 10.

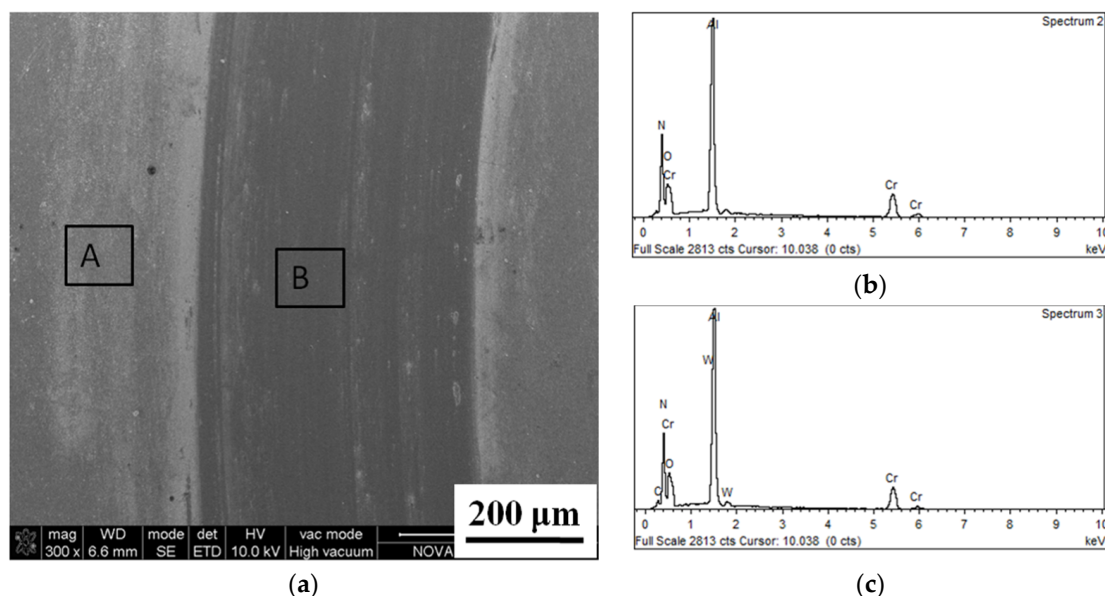


Figure 13. SEM image of wear track (a) and the corresponding energy-dispersive X-ray spectrometer (EDX) results of the intact surface in “A” (b) and the worn surface in “B” (c) on the AlCrN coatings deposited at 450 V charge voltage after the pin-on-disk wear tests at 800 °C.

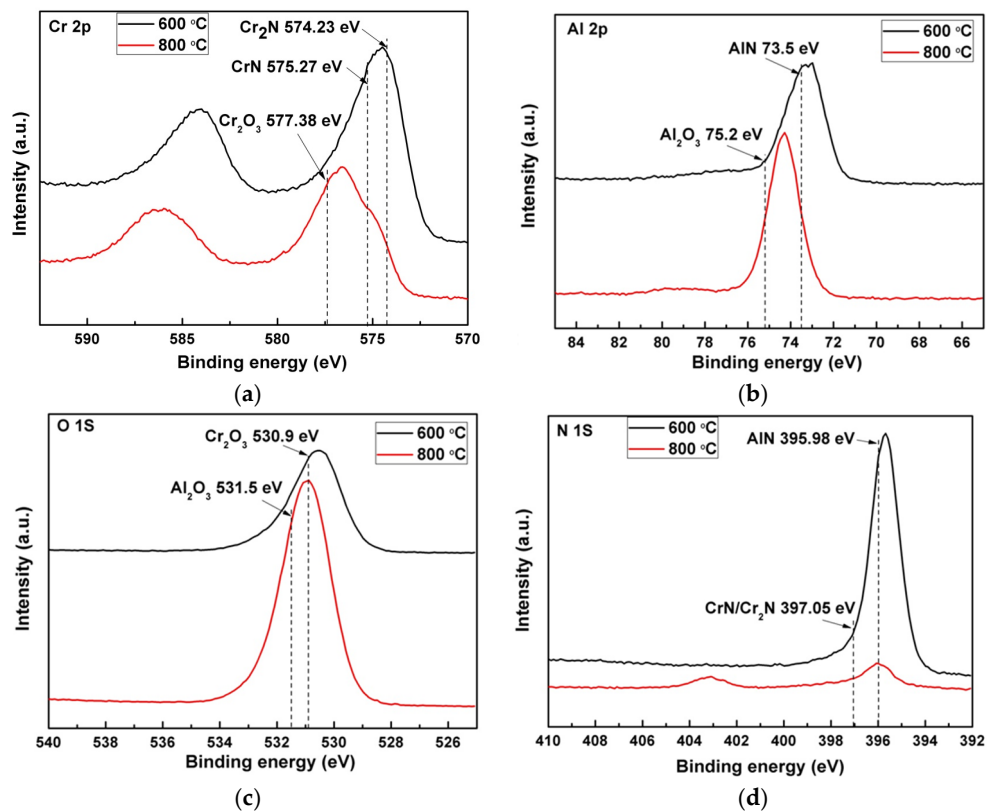


Figure 14. Detailed core level XPS spectra on the worn surface of the AlCrN coatings deposited at 450 V charge voltage after the pin-on-disk wear tests at 600 °C and 800 °C: (a) Cr 2p; (b) Al 2p; (c) O 1s; (d) N 1s.

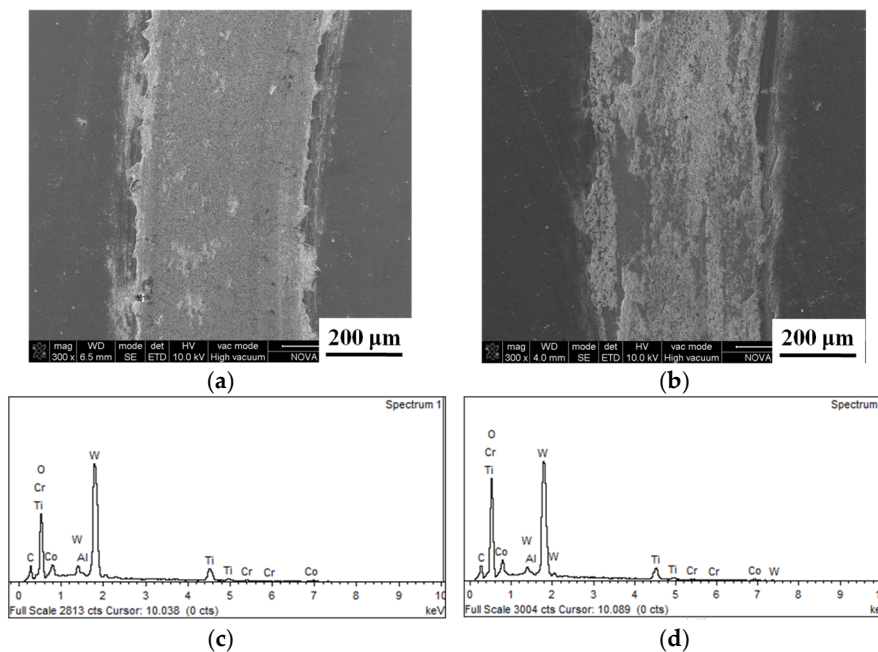


Figure 15. SEM images of wear tracks on the AlCrN coatings deposited at 550 V charge voltage after the pin-on-disk wear tests at (a) 600 °C and (b) 800 °C, (c) and (d) show the EDX results on the wear tracks in (a) and (b), respectively.

4. Conclusions

Five AlCrN coatings were deposited using MPPMS with varying charge voltages applied to the AlCr targets. All the AlCrN coatings exhibited a dense columnar microstructure. The main phase in the AlCrN coatings was the c-(Al, Cr)N solid solution phase with a NaCl-type structure. A h-AlN phase appeared in the coatings when the charge voltage was higher than 500 V. The hardness of the AlCrN coatings was also influenced by charge voltages. The hardness values increased from 24.9 GPa at charge voltage of 350 V to 30.8 GPa at 450 V, and then decreased to 20.6 GPa at 550 V. During vacuum annealing at temperatures below 800 °C, no remarkable difference was observed in the coating structure or in the mechanical properties. When the annealing temperature increased to 1000 °C, grain growth and the formation of h-AlN and Cr₂N phases occurred in the coatings. Correspondingly, the hardness visibly decreased for all the coatings. The friction coefficients and wear rates of the coatings against Al₂O₃ balls exhibited a dependence on both the charge voltage during coating deposition and the temperature of the wear test. The reasons for this are related to the microstructure, phase structure and hardness of the coatings as functions of the charge voltage.

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